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date:_____. CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure. Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger. _____ 1. Greek thinker; called nature ' s basic particle an atom, based on the Greek word " atomos " which means " indivisible " . Did not have evidence that atoms existed.

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Questions : 1. The number of electrons in Na⁺ (atomic number = 11) are (a) 11 (b) 10 (c) 12 (d) 13 2. There are four elements P, Q, R and S having atomic numbers of 4, 18, 10 and 16 respectively. The element which can exhibit covalency as well as electrovalency will be: (a)...

Chapter 4 Structure of Atom MCQ Test 2 Science| Class 9th

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4.05 Atomic Number, Mass number, Isotopes and Isobars. Chapter 4 Structure of Atoms - Test. 05 The Fundamental Unit of Life 5 Topics | 6 Quizzes 5.01 Introduction to cell 5.01 Introduction to Cell. 10.03 Plasma Membrane 5.02 Plasma Membrane. 10.04 Cell wall and Nucleus

Atomic Structure | Class 9 Chapter 4 Structure of Atom ...

The four atomic species can be represented as follows. Out of these, the two species which can be termed isobars are: (i) $^{201}_{60}\text{X}$ (ii) $^{200}_{61}\text{X}$ (iii) $^{200}_{58}\text{X}$ (iv) $^{203}_{60}\text{X}$

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Chemistry Chapter 4 Atomic Structure Transformed

Democritus's ideas of atoms into a scientific theory: 1) All elements are composed of tiny indivisible particles called atoms. 2) Atoms of the same element are identical. 3) Atoms

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of different elements can mix together forming mixtures of whole number ratio compounds. 4)Chemical reactions occur when atoms are seperated, joined or rearranged.

Chemistry Chapter 4 Atomic Structure Test

Weighted Atomic mass is $26.4 + 8.9 = \underline{\hspace{2cm}}$ amu. Title:

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A 5. A 6. A 7. C 8. A 9. B 10. A 11. B 12. A 13. D 14. B 15 ...

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